

AMENDMENTS TO THE CLAIMS**Claims 1 to 64. (Canceled)**

**Claim 65. (New)** A nano-structure of oxide or complex oxide of a metal element, wherein the metal element is at least one selected from the group consisting of transition metal elements, group IA elements, group IIA elements, group IIIB elements, group IVB elements, group VB elements and group VIB elements and has an ability to compose a fluoride complex ion, and

wherein a stability constant of the metal fluoride complex is smaller than that of aluminum fluoride.

**Claim 66. (New)** A nano-structure according to claim 65, wherein an aluminum template can be substituted by said fluoride complex ion.

**Claim 67. (New)** A stacked nano-structure of oxide made from the first oxide or complex oxide of a metal element and the second oxide or complex oxide of a metal element,

wherein the metal element is at least one selected from the group consisting of transition metal elements, group IA elements, group IIA elements, group IIIB elements, group IVB elements, group VB elements and group VIB elements and has an ability to compose a fluoride complex ion, and

wherein a stability constant of the metal fluoride complex is smaller than that of aluminum fluoride.

**Claim 68. (New)** The nano-structure according to claim 65, wherein the oxide or complex oxide comprises fine particles of metal.

**Claim 69. (New)** The nano-structure according to claim 65, wherein aluminum oxide remains in an amount of 0.1 volume% or more, relative to the total oxide.

**Claim 70. (New)** A nano-structure which is made by nitriding, reducing, and carbonizing the nano-structure of oxide according to claim 65.

**Claim 71. (New)** The nano-structure according to claim 65, which is a nano-hole array wherein nano-holes which have penetrating pores of 50  $\mu\text{m}$  or more, are arranged like a bundle.

**Claim 72. (New)** The nano-structure according to claim 71, wherein the aspect ratio is 100 or more.

**Claim 73. (New)** The nano-structure according to claim 65, which is a nano-hole array with a substrate, wherein the nano-holes are arranged like a bundle on at least one main surface of the substrate.

**Claim 74. (New)** The nano-structure according to claim 73, wherein the length of the nano-hole is 1  $\mu\text{m}$  or more.

**Claim 75. (New)** The nano-structure according to claim 73, wherein the aspect ratio is 5 or more.

**Claim 76. (New)** The nano-structure according to claim 73, wherein the substrate is electrically conductive metal or non-metal.

**Claim 77. (New)** The nano-structure according to claim 65, which is a nano-rod of oxide.

**Claim 78. (New)** The nano-structure according to claim 77, wherein the length of the nano-rod is 1  $\mu\text{m}$  or more.

**Claim 79. (New)** The nano-structure according to claim 77, wherein the aspect ratio is 5 or more.

**Claim 80. (New)** The nano-structure according to claim 65, which is a nano-needle of oxide.

**Claim 81. (New)** The nano-structure according to claim 80, wherein the length of the nano-hole is 1  $\mu\text{m}$  or more.

**Claim 82. (New)** The nano-structure according to claim 80, wherein the aspect ratio is 5 or more.

**Claim 83. (New)** The nano-structure according to claim 80, wherein the inside diameter is 10 to 500 nm.

**Claim 84. (New)** A method of preparing a nano-structure of oxide, which comprises:

- a step of preparing a template which has a nano-structure and is made from oxide;
- a step of preparing a solution which contains a fluoride complex ion of the metal element of the target oxide; and
- a step of immersing the oxide template into the solution to substitute the oxide template with the target oxide.

**Claim 85. (New)** The method of preparing a nano-structure of oxide according to claim 84,

- wherein the target oxide is a metal element which is at least one selected from group consisting of transition metal elements, group IA elements, group IIA elements, group IIIB elements, group IVB elements, group VB elements and group VIB elements and has an ability to compose a fluoride complex ion, and

- wherein the stability constant of the fluoride complex is smaller than that of aluminum fluoride.

**Claim 86. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the target oxide is the oxide of the metals, fluoride of which is soluble in the water and can be hydrolyzed, and the fluoride complex ion of which is unstable than the aluminum fluoride.

**Claim 87. (New)** The method of preparing a nano-structure of oxide with a substrate according to claim 84, wherein the template is made from oxide and has a layer having nano-structure provided on at least one main surface of the substrate.

**Claim 88. (New)** The method of preparing a nano-structure of oxide with a substrate according to claim 84, wherein the substrate is metal or non-metal.

**Claim 89. (New)** The method of preparing a nano-structure of oxide with a substrate according to claim 84, wherein the template is used which has a layer of aluminum oxide having a nano-structure formed by anodization treatment (anodized alumina) on at least one main surface of an aluminum metal substrate.

**Claim 90. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the fluoride complex ion is in an aqueous solution at a concentration of 0.1 mmol/l or more.

**Claim 91. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the fluoride complex ion is prepared in which the fluoride complex is present in the form of  $\text{MF}_x^{\text{Y}-}$  (wherein M is a transition metal element, a group IA element, a group IIA element, a group IIIB element, a group IVB element, a group VB element or a group VIB element, x is the number of fluorine atoms and y is an valency).

**Claim 92. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the target oxide is formed via a hydroxide which is formed by hydrolysis of the fluoride complex ion in the solution.

**Claim 93. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the substitution reaction between the oxide of the template and the target oxide is carried out by a dissolution reaction of the oxide of the template and a precipitation reaction of the target oxide.

**Claim 94. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the substitution reaction is carried out in the range of 0 to 80°C under atmospheric pressure.

**Claim 95. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the substitution reaction is carried out in the range of 5 to 40°C under atmospheric pressure.

**Claim 96. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the substitution reaction comprises at least a first substitution reaction which is conducted in a solution comprising the first fluoride complex ion, and a second substitution reaction which is conducted in a solution comprising the second fluoride complex ion, which reactions are sequentially conducted, to prepare a nano-hole array of oxide wherein at least the first metal oxide and the second metal oxide are stacked.

**Claim 97. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the substitution reaction comprises a substitution reaction which is carried out in a solution comprising at least the first fluoride complex ion and the second fluoride complex ion, to prepare a nano-hole array of oxide comprising a complex oxide of at least the first metal oxide and the second metal oxide.

**Claim 98. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the substitution reaction comprises a substitution reaction which is carried out in a solution comprising at least one kind of fluoride complex ion and at least

one kind of fine metal particles, to prepare a nano-hole structure of oxide comprising the fine metal particles.

**Claim 99. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the substitution reaction is carried out under any of light irradiation, radioactive ray irradiation and ultrasonic irradiation.

**Claim 100. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the template is used which comprises aluminum oxide having a nano-structure formed by anodization treatment (anodized alumina).

**Claim 101. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the template is used which has a structure in which pores are regularly extended on one surface.

**Claim 102. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the template is used which has a structure in which pores penetrates from one surface to the other surface.

**Claim 103. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the template is used which has a structure having pores of 200 nm diameter on one surface and having pores of 20 nm diameter on the other surface.

**Claim 104. (New)** The method of preparing a nano-structure of oxide according to claim 84,

wherein the nano-structure is in the form of a nano-rod, and

wherein the substitution process is a reaction of substituting the oxide of the template with the target oxide by making the precipitation reaction rate of the target metal oxide greater than the dissolution reaction rate of anodized alumina.

**Claim 105. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the substitution reaction is carried out in the range of 20 to 80°C under atmospheric pressure.

**Claim 106. (New)** The method of preparing a nano-structure of oxide according to claim 84, wherein the substitution reaction is carried out under addition of a fluoride ion scavenger.

**Claim 107. (New)** The method of preparing a nano-needle of oxide according to claim 84, which comprises a step of separating the nano-hole array of oxide into each of nano-holes of oxide (nano-needles).

**Claim 108. (New)** A high-performance nano-hole array, which is a nano-hole array made from oxide or complex oxide of a metal element,

wherein the metal element is at least one selected from the group consisting of transition metal elements, group IA elements, group IIA elements, group IIIB elements, group IVB elements, group VB elements and group VIB elements and has an ability to compose a fluoride complex ion,

wherein the stability constant of the fluoride complex is smaller than that of aluminum fluoride, and

wherein the penetrating pores of the nano-holes, which have the length of 50  $\mu\text{m}$  or more and the aspect ratio of 100 or more, are arranged like a bundle, or the nano-holes, which have bottoms and have the length of 1  $\mu\text{m}$  or more and the aspect ratio of 5 or more, are arranged like a bundle on at least one main surface of the substrate.

**Claim 109. (New)** The high-performance nano-hole array according to claim 108 responsive to visible light,

wherein the oxide is  $\text{TiO}_2$ ,  $\text{ZnO}$ ,  $\text{SnO}_2$ ,  $\text{SiO}_2$  or a mixture thereof, or a complex oxide thereof, and

wherein at least one selected from the group consisting of Ag, Pt and Cu fine particles is dispersed.

**Claim 110. (New)** The nano-hole array according to claim 108 for photochromism, wherein the oxide is  $\text{TiO}_2$  or  $\text{SiO}_2$ , and Ag is supported.

**Claim 111. (New)** The nano-hole array according to claim 108 for an energy-saving photocatalyst, wherein  $\text{WO}_3$  is supported in the nano-hole.

**Claim 112. (New)** The nano-hole array according to claim 108 which is used for contacting the electrolyte in a dye sensitization type of a solar cell.

**Claim 113. (New)** The nano-hole array according to claim 108 for a positive electrode of a lithium-ion battery, wherein the oxide is  $\text{V}_2\text{O}_5$  or  $\text{TiO}_2$ .

**Claim 114. (New)** The nano-hole array according to claim 108 for a material for thermoelectric conversion, wherein the oxide is ZnO or TiO.

**Claim 115. (New)** The nano-hole array according to claim 108 for a material for thermoelectric conversion, wherein the oxide is ZnO,  $\text{TiO}_2$ ,  $\text{SnO}_2$ ,  $\text{Fe}_2\text{O}_3$  or  $\text{ZrO}_2$  and the nano-metal is embedded in the nano-hole.

**Claim 116. (New)** The nano-hole array according to claim 108 for a gas sensor wherein the oxide is TiO,  $\text{TiO}_2$ , ZnO,  $\text{SnO}_2$  or a mixture thereof, or a complex oxide thereof.

**Claim 117. (New)** The nano-hole array according to claim 108 for a humidity sensor, wherein the oxide is  $\text{SnO}_2$ .

**Claim 118. (New)** The nano-hole array according to claim 108 for an odor sensor, wherein the oxide is TiO,  $\text{TiO}_2$ , ZnO,  $\text{SnO}_2$  or a mixture thereof, or a complex oxide thereof.



**Claim 119. (New)** The nano-hole array according to claim 108 for a light sensor or a photonic crystal, wherein the oxide is  $\text{TiO}_2$ .

**Claim 120. (New)** The nano-hole array according to claim 108 for a filter, wherein the oxide is oxide other than  $\text{Al}_2\text{O}_3$ .

**Claim 121. (New)** The nano-hole array according to claim 108 for a material for  $\text{CO}_2$  mobilization, wherein the oxide is represented by a formula  $\text{MO}_b$  (wherein M is Zr, Fe, Ni, Ti or Si and b is the number of oxygen atoms) or a formula  $\text{Li}_a\text{MO}_b$  (wherein M is Zr, Fe, Ni, Ti or Si, a is the number of lithium atoms, and b is the number of oxygen atoms).

**Claim 122. (New)** The nano-hole array according to claim 108 for high-density memory media, wherein the oxide is a stacked oxide comprising any one of the combinations of  $\text{Fe}_2\text{O}_3$  and  $\text{ZrO}_2$ ,  $\text{Fe}_2\text{O}_3$  and  $\text{TiO}_2$ ,  $\text{Fe}_2\text{O}_3$  and  $\text{SnO}_2$ ,  $\text{Fe}_3\text{O}_4$  and  $\text{ZrO}_2$ ,  $\text{Fe}_3\text{O}_4$  and  $\text{TiO}_2$ , and  $\text{Fe}_3\text{O}_4$  and  $\text{SnO}_2$ .

**Claim 123. (New)** A nano-rod, which is separated, respectively, made from oxide or complex oxide of a metal element,

wherein the metal element is at least one selected from the group consisting of transition metal elements, group IA elements, group IIA elements, group IIIB elements, group IVB elements, group VB elements and group VIB elements and has an ability to compose a fluoride complex ion,

wherein the stability constant of the fluoride complex is smaller than that of aluminum fluoride, and

wherein the length of the nano-rod is 1  $\mu\text{m}$  or more and the aspect ratio of the nano-rod is 5 or more.

**Claim 124. (New)** The nano-rod according to claim 123 for a material for matrix reinforcement, wherein the oxide is  $\text{TiO}_2$ ,  $\text{ZnO}$ ,  $\text{SnO}_2$ ,  $\text{SiO}_2$  or a mixture thereof, or a complex oxide thereof.

**Claim 125. (New)** The nano-rod according to claim 123 for a photocatalyst, wherein the oxide is  $\text{TiO}_2$ ,  $\text{ZnO}$ ,  $\text{SnO}_2$ ,  $\text{SiO}_2$  or a mixture thereof, or a complex oxide thereof.

**Claim 126. (New)** A nano-needle for micro-injection, which is separated, respectively, made from oxide or complex oxide of a metal element, wherein the metal element is at least one selected from the group consisting of transition metal elements, group IA elements, group IIA elements, group IIIB elements, group IVB elements, group VB elements and group VIB elements and has an ability to compose a fluoride complex ion and the stability constant of the fluoride complex is smaller than that of aluminum fluoride, and wherein the length of the nano-needle is 1  $\mu\text{m}$  or more and the aspect ratio is 5 or more.

**Claim 127. (New)** The nano-needle for micro-injection according to claim 126, wherein the oxide is  $\text{ZnO}$ ,  $\text{TiO}_2$  or  $\text{SnO}_2$ .

**Claim 128. (New)** The nano-needle for micro-operation according to claim 126, wherein the oxide is  $\text{ZnO}$ ,  $\text{TiO}_2$  or  $\text{SnO}_2$ .

**Claim 129. (New)** The nano-needle for micro-adhesion according to claim 126, wherein the oxide is  $\text{ZnO}$ ,  $\text{TiO}_2$  or  $\text{SnO}_2$ .